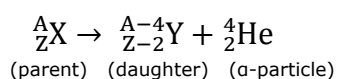


Experiment Name: Radioactive decay law

Aim: To draw radioactivity graph

Theory:

A nucleus of a radioactive element that emits an α -particle must transform into a nucleus of another element. The nucleus of the so-called 'parent' element loses two neutrons and two protons. Therefore the nucleon number (A) changes by 4 and the proton number (Z) changes by 2. The nucleus formed by this decay is called the 'daughter nucleus'. We may express such a nuclear decay by the nuclear reaction equation



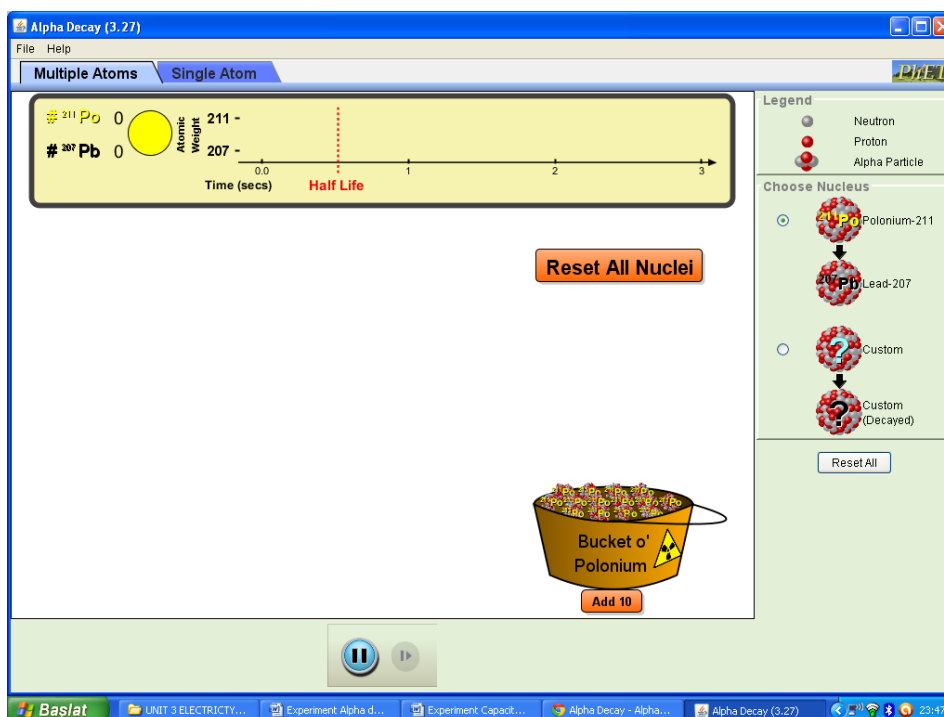
The radioactive decay law enables us to determine a relation between the half-life of a radioactive element and the decay constant.

If a sample of a radioactive element initially contains N_0 atoms, after an interval of one half-life the sample will contain N atoms. If the half-life of the element is $T_{1/2}$ from the decay law, we can write that

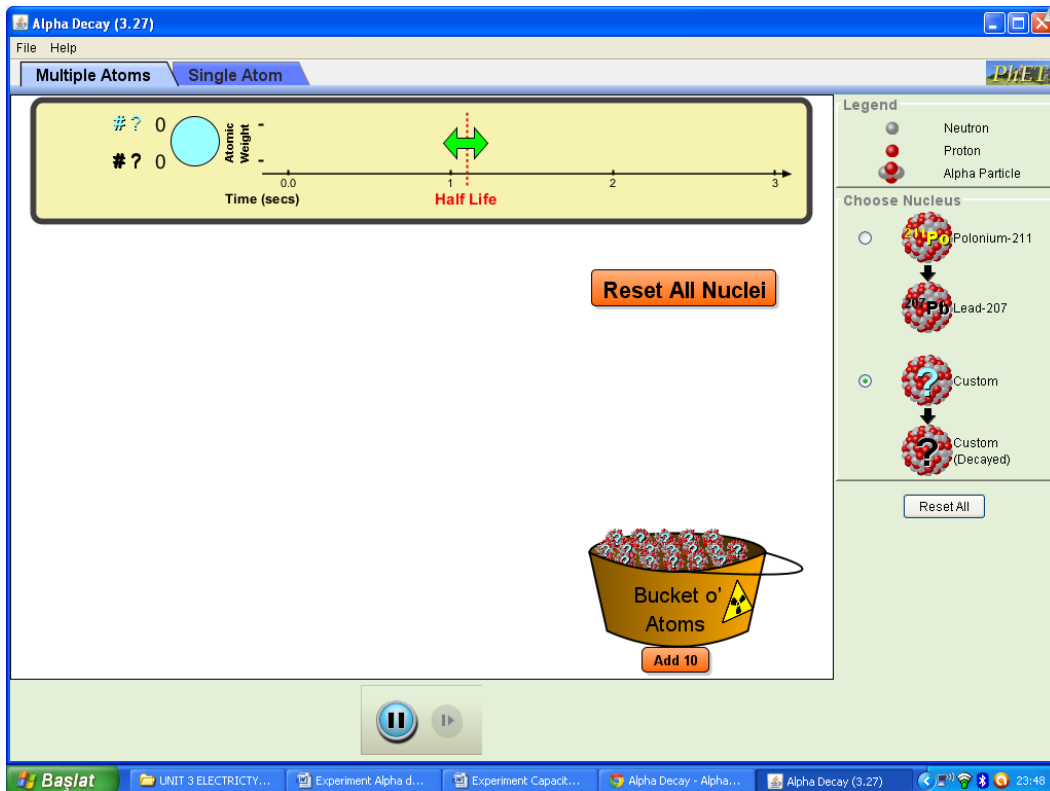
$$N = N_0 e^{-\lambda t}$$

Procedure:

- Go to <http://phet.colorado.edu/en/simulation/alpha-decay>, and click on the "Run Now!" button. The below window should appear.

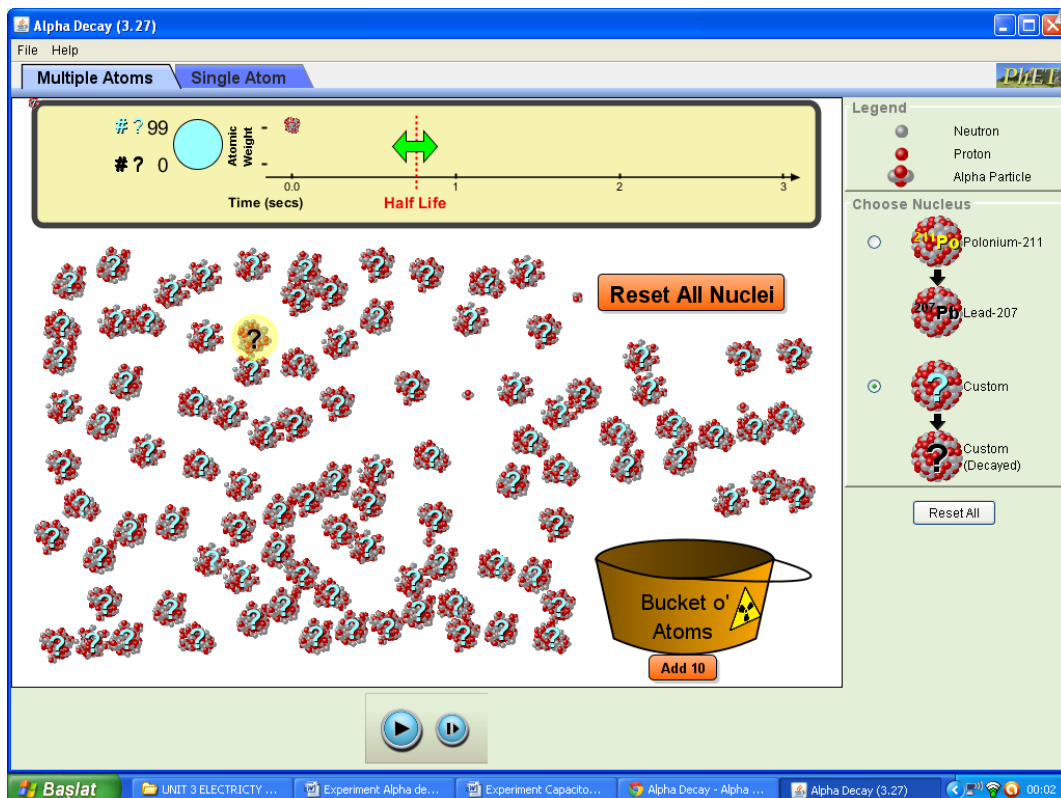


2. Click to “Custom” from the column on the right of the screen and get the below diagram.



3. Adjust the half life to any value you want between 0.5-1sec. using double-sided green arrow.

4. Add atoms to the bucket by clicking “Add 10” icon below the bucket at the bottom and get the below diagram.



5. By clicking “▶” button, get how many undecayed atom remains in each time interval of 0.5s.
Repeat your measurement 10 times, write your data to the below table and calculate the average.

Time (s)	0	0.5	1.0	1.5	2.0	2.5	3.0
Number of undecayed atom, N	99						
	99						
	99						
	99						
	99						
	99						
	99						
	99						
	99						
	99						
Average	99						

6. Draw number of undecayed atom vs time graph.
7. Calculate values of **ln N** (average) and write them in the table below.

Time (s)	0	0.5	1.0	1.5	2.0	2.5	3.0
Number of undecayed atom, N	99						
ln N							

8. Draw ln N vs time graph and calculate its slope which is known as decay constant, λ .

$$\lambda = \frac{\ln N}{T_{1/2}}$$

9. Using the below relation, calculate half life of parent nucleus.

$$T_{1/2} = \frac{\ln N}{\lambda}$$